

lecture 21.10.2010

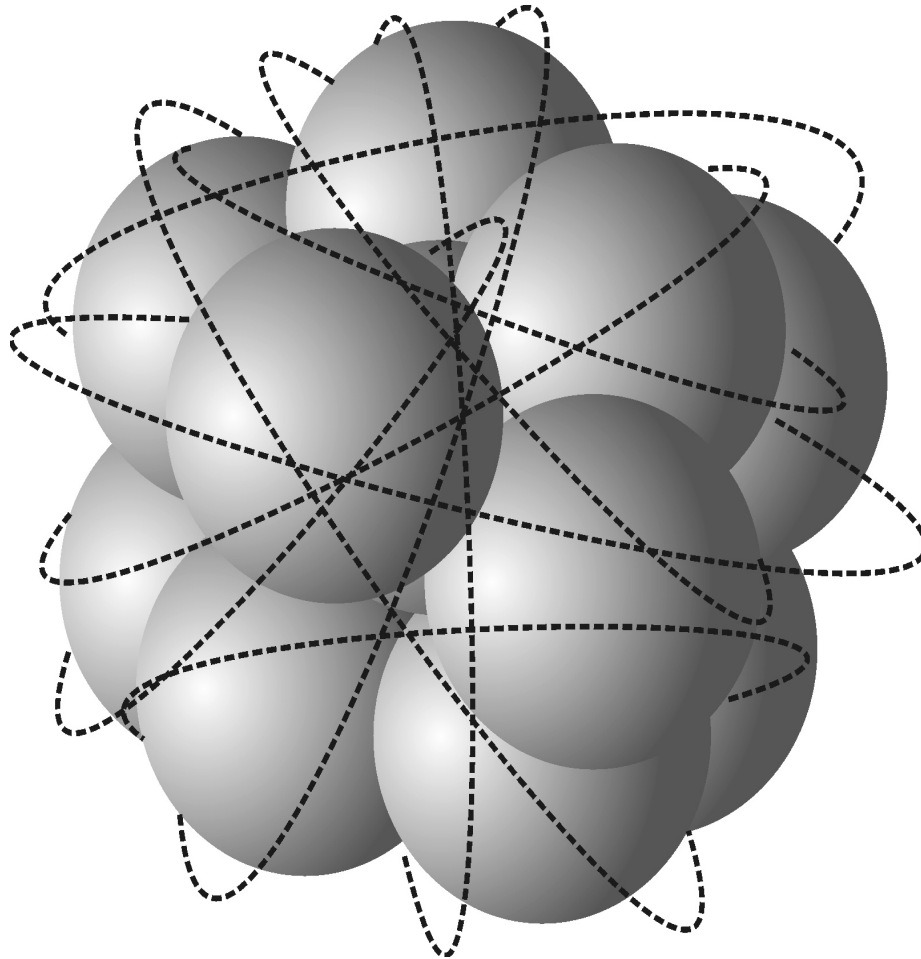
we had in the last week:

- introduction into atomic physics, relevance
- introduction into cluster physics

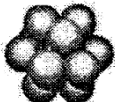
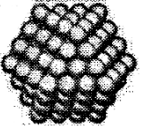
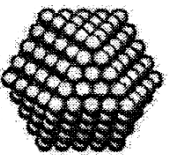
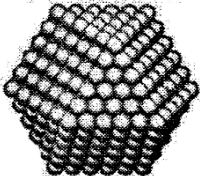
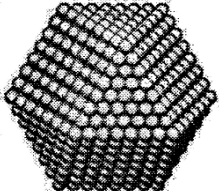
today:

- few more remarks on cluster physics
- atomic physics: history
- preparation of atomic beams
- selected examples with ultracold beams

A cluster of simple metal atoms can be considered as a metallic quantum dot. In the corresponding bulk materials the electrons need much more space.



size-dependence

Full-shell Clusters		Total Number of Atoms	Surface Atoms (%)
1. Shell		13	92
2 Shells		55	76
3 Shells		147	63
4 Shells		309	52
5 Shells		561	45
7 Shells		1415	35

Electronic shell structure of metal clusters

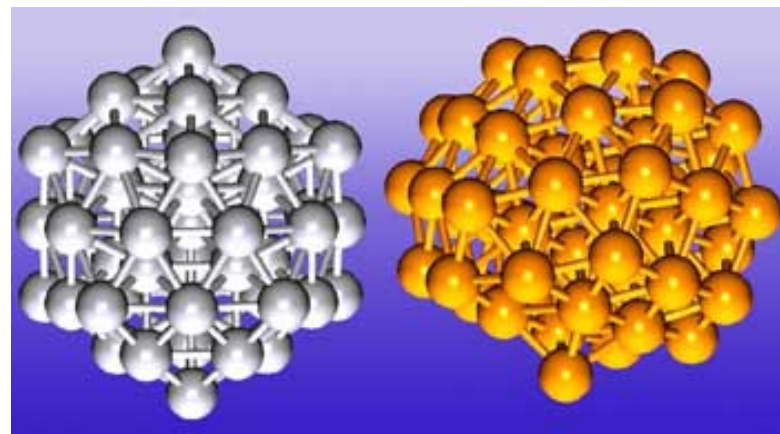
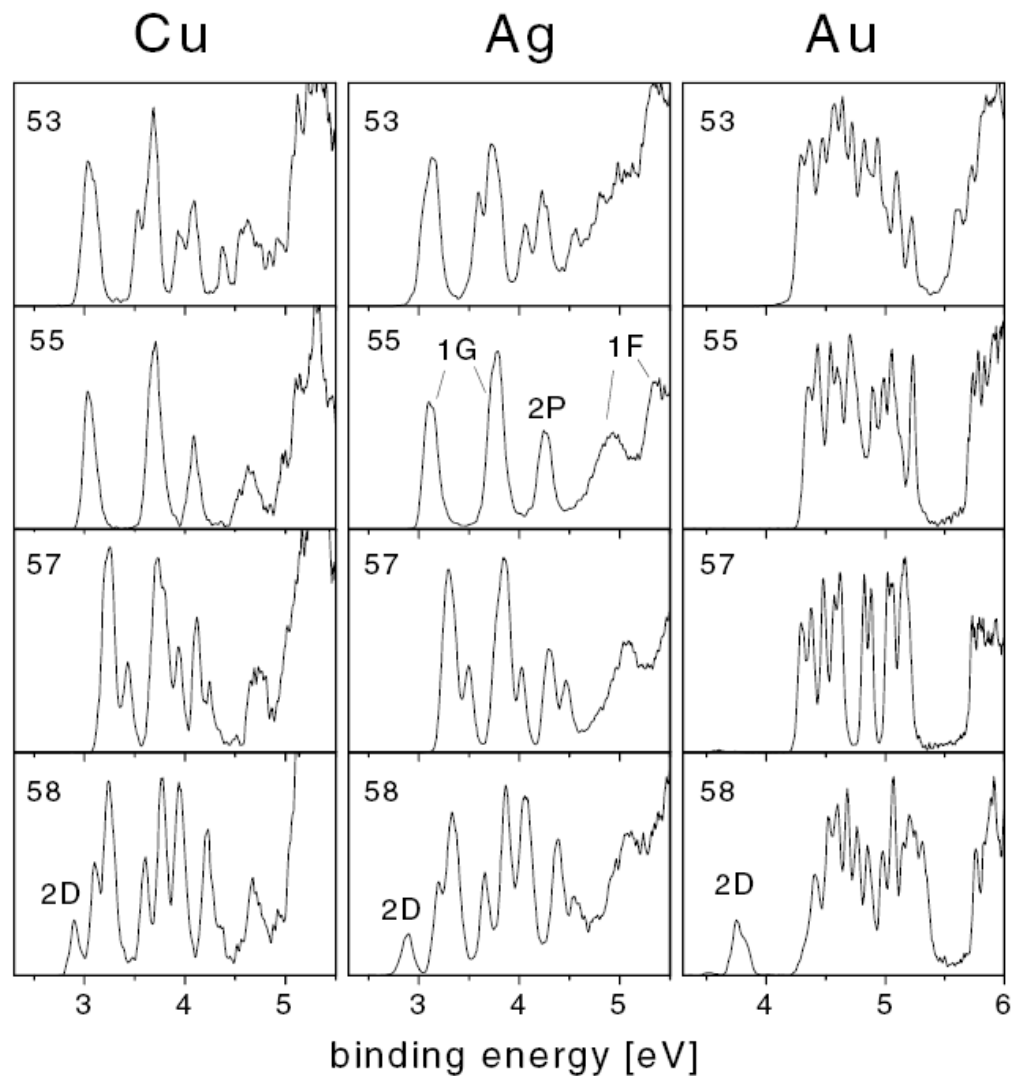
	5s (2)	330	Al ₁₁₀
	4d (10)	328	
	3g (18)	318	Al ₁₀₆
	2i (26)	300	Al ₁₀₀
1k (34)		274	Al ₉₁ ⁻
	4p (6)	240	Al ₈₀
	3f (14)	234	Al ₇₈
	2h (22)	220	Al ₇₃ ⁻
1j (30)		198	Al ₆₆
	4s (2)	168	Al ₅₆
	3d (10)	166	Al ₅₅ ⁻
	2g (18)	156	Al ₅₂
1i (26)		138	Al ₄₆
	3p (6)	112	Al ₃₇ ⁻
	2f (14)	106	Al ₃₅ ⁻
1h (22)		92	
	3s (2)	70	Al ₂₃ ⁻
	2d (10)	68	
1g (18)		58	Al ₁₉ ⁻
	2p (6)	40	Al ₁₃ ⁻
1f (14)		34	Al ₁₁ ⁻
	2s (2)	20	
1d (10)		18	
1p (6)		8	
1s (2)		2	

example: Al_N
3 electrons per atom

will be derived later
in this lecture

FIG. 1. Spherical shell closings and the corresponding closed shell neutral and negative clusters of trivalent Al. The number in the parentheses after each shell index indicates the occupation number, $2(2l + 1)$.

photoelectron spectroscopy



Hakkinen et al., PRL 93
093401(2004)

FIG. 1. Photoelectron spectra of Cu_n^- , Ag_n^- , and Au_n^- ($n = 53, 55, 57, 58$) obtained at a photon energy of 6.42 eV.

beginning of atomic physics:

- Leuklipp 440 v Chr, Demokrit 460-370 v Chr
ατομος 'completely undivisible' (gänzlich unteilbar).
idea: outside the atom empty space
- Epikur 341-271 v Chr: atom -- gravity (Schwere)
- later the idea of 'atom' has been forgotten for centuries
- for the church the reinvention of the atom has been too materialistic;
in 1348 Nicolas von Antrecourt had to retract (widerrufen) his ideas
on the atomic nature
- break-through from the 17th century on mainly by chemists
(weighing, reactants and products)
- 19. Jht. Clausius, Maxwell, Boltzmann: p, T, specific result from
atomic collisions

few historical key experiments

modern Millikan experiment with magnetic levitation

Electric Neutrality of Matter

G. Gallinaro, M. Marinelli, and G. Morpurgo

Istituto di Fisica, Università di Genova, Genova, Italy, and Istituto Nazionale di Fisica Nucleare –

Sezione di Genova, Italy

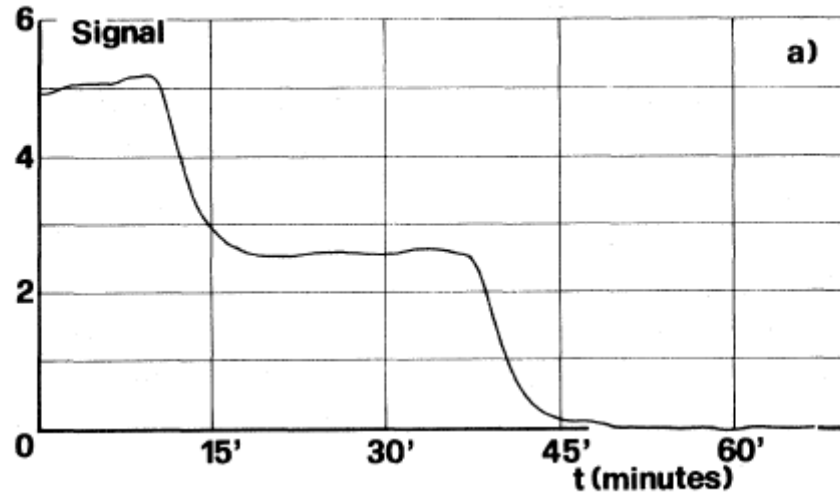
(Received 18 April 1977)

With the use of a new feedback levitation electrometer (with an increase in sensitivity by 10^3 in comparison to our previous graphite experiments) iron objects of mass $\sim 2 \times 10^{-4}$ g have been explored for fractionally charged quarks and/or a possible electron-proton charge difference. Upper limits found were $N(\text{quarks})/N(\text{nucleons}) < 3 \times 10^{-21}$ and $(Q_p - |Q_e|)/Q_p < 10^{-21}$. The present "sensitivity" is $\sim 10^7$ times that of the original Millikan experiment.

VOLUME 38, NUMBER 22

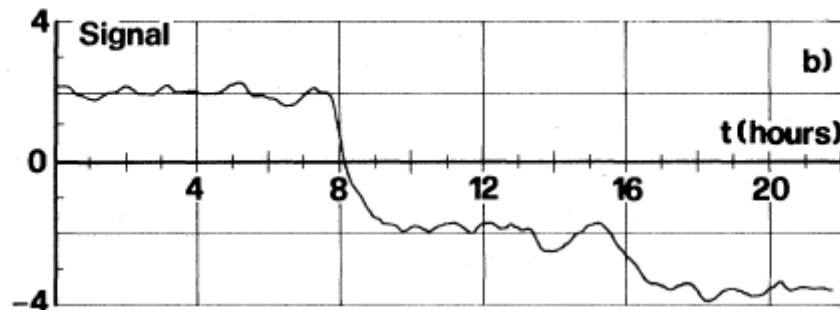
PHYSICAL REVIEW LETTERS

30 MAY 1977



a) charges in the charge of a steel sphere with 3.3×10^{-5} g
Residual charge smaller than $1/40 e$

b) rotating cylinder with 2.5×10^{-4} g.
Charge changes from +1 to -1 and -2



Results:

- upper limit $N(\text{quarks})/N(\text{Nucleons})$ smaller than 3×10^{-21}
- $(Q_p - Q_e)/Q_p$ smaller than 10^{-21}
- sensitivity 10^7 times Millikan

Basics of atom (and molecular) spectroscopy - how to overcome effects of motion and collisions

**one aim: to do undisturbed spectroscopy to learn, e.g.,
about line widths (lecture tomorrow by Prof. Bauer)**

Doppler-reduced spectroscopy: blackboard script

Introduction to supersonic beam expansion: blackboard script

preparation of ultracold atomic beams by supersonic jet expansion

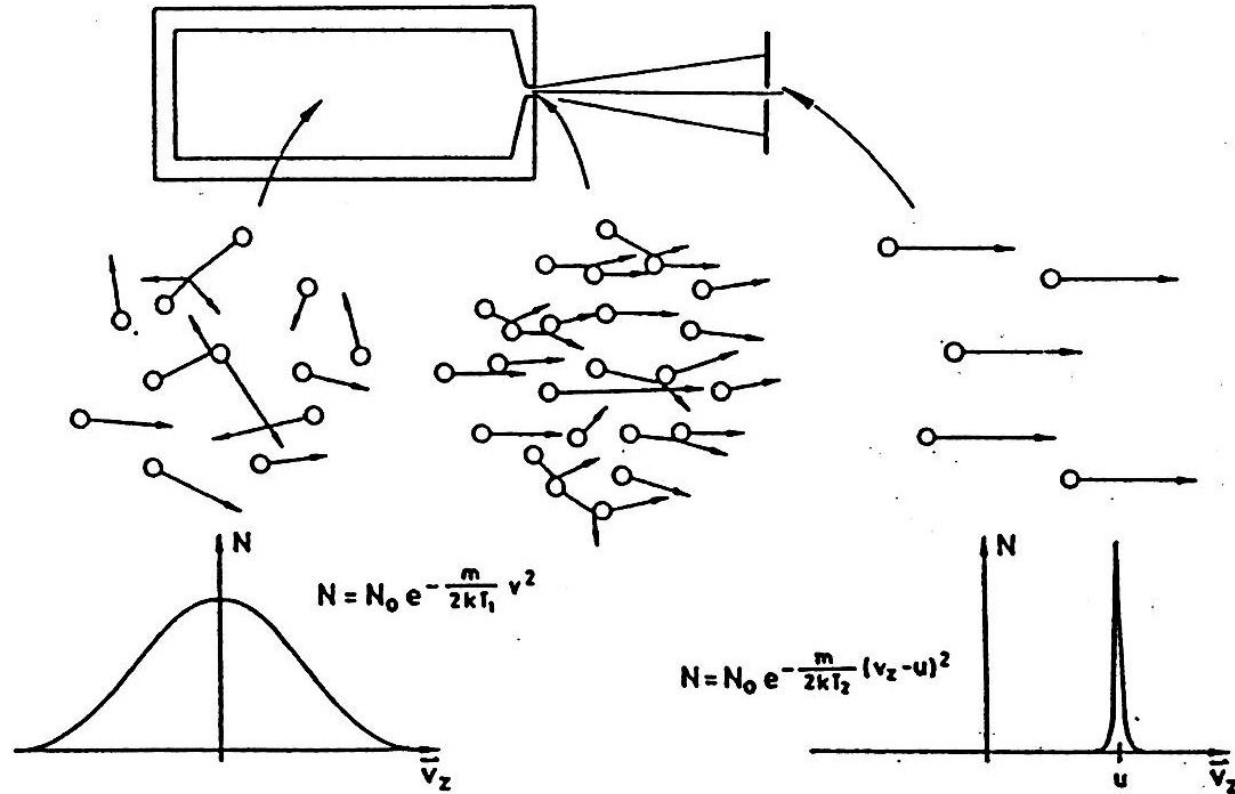


Fig. 8: Die mikroskopische Ursache für die Temperaturniedrigung und Clusterbildung bei der adiabatischen Expansion [13]. Im Ausgangsbehälter herrscht vor der Expansion thermisches Gleichgewicht. Die ungerichtete Geschwindigkeitsverteilung der Teilchen ist durch die Maxwellverteilung bestimmt. Bei der Expansion des Gases durch eine Düse kommt es durch Stöße mit den anderen Gasparkeln zur Abnahme der inneren Energie zu Gunsten der Expansionsgeschwindigkeit u . Die Relativgeschwindigkeit im kalten Molekularstrahl nimmt stark ab, wodurch die "kalten" Atome zu Clustern zusammenlagern.